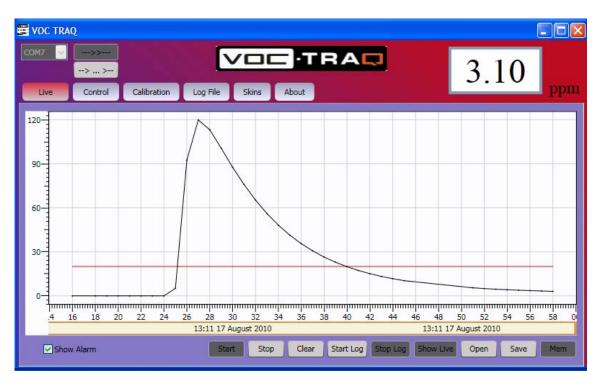
Introduction



This Manual will familiarize the customer with the Baseline MOCON VOC-TRAQ using the piD-TECH® **plus** photo ionization sensor. This includes the principle of operation, technical characteristics, as well as some PID-specific application features. The Manual will instruct customers to easily operate the VOC-TRAQ and the TRAQ-WARE software.



Section 1 1.1 VOC-TRAQ Overview



Portable Photoionization Detector (PID)

The VOC-TRAQ total volatile organic compound (TVOC) detector is an excellent way to monitor and record TVOC's. It operates using any PC with a Windows[®] operating system. Using the award winning Baseline[®]-MOCON[®] piD-TECH[®] plus photoionization detector makes the VOC-TRAQ the most reliable, accurate, and inexpensive portable TVOC detector on the market.

Designed for ease of use all air quality consultants, safety engineers, maintenance managers, or anyone concerned about TVOC's in the indoor environment.

The USB compatible VOC-TRAQ can operate while connected to a PC or it can remotely store up to 36,000 sample readings with the detector's internal memory using the optional rechargeable power supply. Alarm levels are programmable with LED and/or audible notification. The VOC-TRAQ is also compatible with numerous USB accessory options.

The VOC-TRAQ is a simple compact design for monitoring TVOC's in most environments. The included software allows easy calibration, setup, display and analysis.

The VOC-TRAQ consists of a cylindrical housing equipped with a USB port and a 3 color LED. On the top, there is are slotted openings that serve as an entrance for analyzed gas

The photo ionization detector and associated electronic circuits are located inside the housing.

The LED indicates:

- Slow (~ once every 5 sec) Blinking Green = power on and operating.
- Blinking Red = Alarm level exceeded. The greater the concentration exceeds the alarm level the faster the flashing.
- Blinking Yellow = Alarm event occurred then returned below alarm level.

The USB port provides:

• USB 5 VDC power for operation (PC, wall connected USB power supply, or optional battery pack).

VOC TRAQ		
COM7 · ···>··· > > Live Control Calibration	Log File Skins About	0.20 _{ppm}
	149 50	51 53
Show Alarm	14:00 17 August 2010 Start Stop Clear Start Log Stop Log Show	Live Open Save Mem

• USB to PC communications to TRAQ-WARE.

Refer to Appendix for serviceable items.

Section 2 VOC-TRAQ Installation, Calibration & Operation

2.1 Installing VOC-TRAQ Software

Insert the VOC-TRAQ Software Installation disk and follow the instructions. When complete the VOC-TRAQ drivers and the TRAQ-WARE software will be installed and ready to use. Be sure to restart your computer after installation.

2.2 Powering Up the VOC-TRAQ

Connect the USB cable to your computer and then connect the VOC-TRAQ. Note that the VOC-TRAQ goes through a initialization sequence upon power up (flashing red LED with beep, flash amber, flash green, then beep).



2.3 VOC-TRAQ Connection & Control

🗮 VOC TRAQ			
COM7> > > Live Control	Calibration Log File Skins		0.01 _{ppm}
Period, sec.:	1.26 1 Set	Alarm Level, ppm: Audio Alarm:	20.00 O on ⊙ off Set
Memory Used, %:	53% Clean	Reset Alarm	Reset
Date, Time	Set	Sensor Type:	Bronze Set
		Get Parameters	

2.3.1 Start: Open the TRAQ-WARE and select the "Control" tab. The display above will appear.

From the COM tab select the VOC-TRAQ COM Port then "Connect". Note: If you are not sure what COM port is the VOC-TRAQ check in Windows "Device Manager" and look under "Ports". The VOC-TRAQ is identified as the "USB Serial Port".

2.3.2 Set Time: Select "Set" Date, Time. Your PC's date and time will be set into the VOC-TRAQ.

2.3.3 Get Parameters: Select the "Get Parameters" tab and the data from the VOC-TRAQ will be displayed.

2.3.4 Parameters:

Version: displays level of firmware on VOC-TRAQ

Period, **Sec**: displays the time between data samples. Any integer can be entered to set the data sample time in seconds. Press "Set" after entering.

--->>---

Start Delay, min: displays the time delay in minutes from power up until data sampling begins. Any integer can be entered to set the sample delay time in minutes. Press "Set" after entering. Zero delay is normal.

Memory Used, **%**: displays the amount of VOC-TRAQ on board memory is used. To clear memory select "Clean".

Alarm Level, ppm: displays the Alarm Level set. Enter the alarm level desired and press set.

The LED alarm indication:

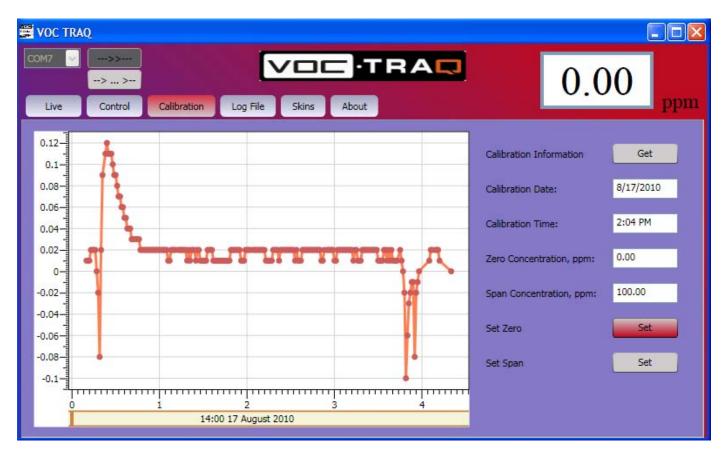
- Slow (~ once every 5 sec) Blinking Green = power on and operating No Alarm.
- Blinking Red = Alarm level exceeded. The greater the concentration exceeds the alarm level the faster the flashing.

Blinking Yellow = Alarm event occurred then returned below alarm level.
 Audible Alarm: Set "on" or "off" to turn on or off the audible beep with alarm.
 Reset Alarm: Press reset to turn off amber "alarm occurred" LED on VOC-TRAQ
 Sensor Type: If the PID sensor is changed the sensor type can be selected.

2.4 Zero Calibration

The VOC-TRAQ should be allowed to stabilize for 15 minutes to an hour before performing a calibration.

As a rule, calibration of the sensor on a daily or weekly basis is recommended. However, if the sensor is used in a relatively clean environment, the calibration frequency can be longer.



- **2.4.1** Select the "Calibration" Tab.
- 2.4.2 By Calibration Information select "Get"
- **2.4.3** In the box next to "Zero Concentration, ppm:" enter the ppm value of your zero gas (usually 0.00).
- **2.4.4** Using the calibration cap apply zero gas to the VOC-TRAQ. A stabilization period of one minute or more should be allowed for zero gas and span gas when they are applied to the sensor during a calibration.
- 2.4.5 Select the "Set" button next to Set Zero to calibrate the zero value.

When selecting the Calibration function the program will automatically switch to a 1 second data collection interval for easier calibration. It will switch back when the Calibration function is exited.

2.5 Span Calibration

🖉 VOC TRAQ		
COM7 V> > > Live Control Calibration Log File Skins		00 _{ppm}
100-	Calibration Information	Get
90	Calibration Date:	8/17/2010
70-	Calibration Time:	2:08 PM
50	Zero Concentration, pr Span Concentration, p	
30-	Set Zero	Set
	Set Span	Set
7	7 August 2010	

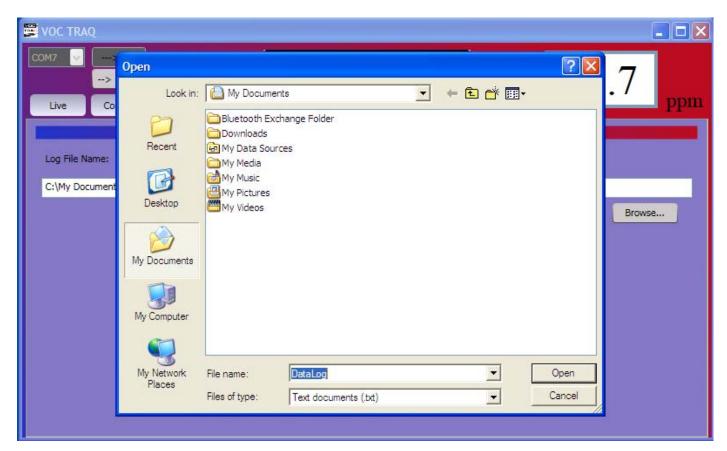
- **2.5.1** In the box next to "Span Concentration, ppm:" enter the ppm value of your span gas.
- **2.5.2** Using the calibration cap apply span gas to the VOC-TRAQ. A stabilization period of one minute or more should be allowed for zero gas and span gas when they are applied to the sensor during a calibration.
- 2.5.3 Select the "Set" button next to Set Span to calibrate the Span value.

This completes the Calibration.

2.6 Log File

📴 VOC TRAQ	
COM7 > > Live Control Calibration Log File Skins About	76.7 _{ppm}
Log File Name: C:\My Documents\VOC_TRAQ_Log.txt	
	Browse

Log File: Here data files can be "Browsed", created and opened. Enter any file name & open.



2.7 Operation

	->>						TF	RA						
	Control	Calibratio	n L	og File	Skins	Abou	ut							1
		_										_	_	
												_	_	
					_		-					_		
					6					12	13	14	15	m
-	-			00:00 1 J				-	 					

2.6.1 Live: Select "Live" Tab to show the display above.

2.6.2 Show Alarm: Select the "Show Alarm" box to display the alarm level on the graph.

2.6.3 Bottom Button Selections:

Start: Starts the VOC-TRAQ data display (default).

Stop: Stops the VOC-TRAQ data display.

Clear: clears the data shown on the graph.

Start Log: Starts the VOC-TRAQ on board data logging.

Stop Log: Stops the data logging.

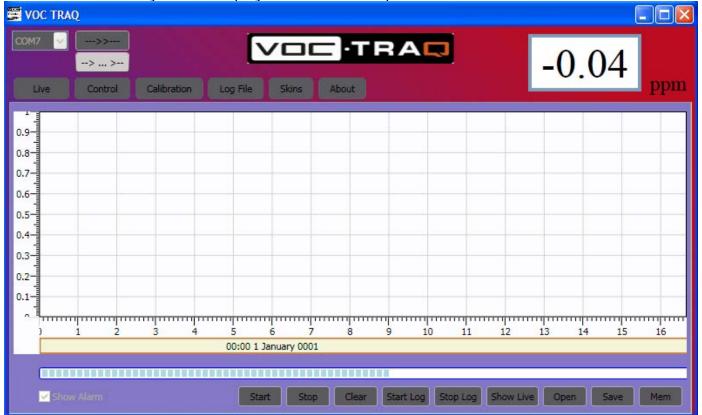
Show Live: Return to actual graph after browsing data.

Open: Opens a data file. There will be a prompt to choose a file name and location.

🖉 VOC TR/	Open					? 🛛		_ 🗆 🗙
СОМ7 🖂	Look in:	My Documents	s	-] + E 💣 🎟	÷		
Live	Recent Desktop My Documents My Computer	Bluetooth Excha Downloads My Data Source My Media My Music My Pictures My Videos DataLog1.txt					5.14	ppm
5	My Network Places	File name: Files of type:	Text document	s (bt)	- -	Open Cancel		
02	15,41		1			August 2010		40 42
Show		17 August 2010	Start St	top Clear	Start Log Stop Log	7 August 2010	Open Save	Mem

Save: Saves the data displayed. There will be a prompt to choose a file name and location. This can save actual data as well as "Mem" data.

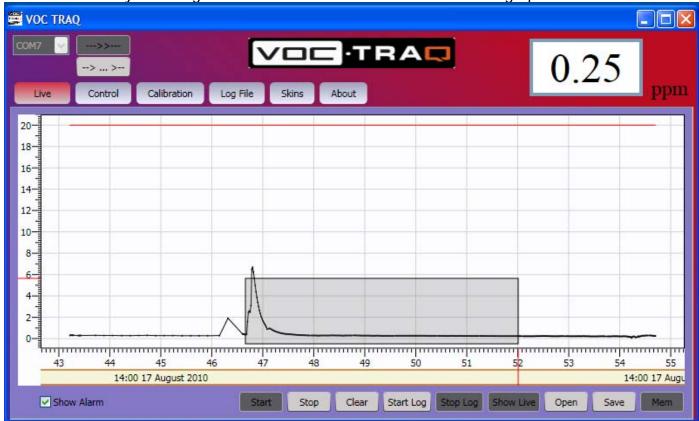
Mem: This tab is accessible once the "Stop" button is selected. The VOC-TRAQ memory can be displayed. It can take up to 4 minutes to download



In the picture above a bar graph count down indicates the download % completion. The internal memory provides more than a week of data - data interval set to 20 sec.

2.8 Dynamic Data Display

Hold the "ctrl" key and drag the mouse to enclose a desired section of graph.



📴 VOC TRAQ		_ 🗆 🛛
COM7 V> > > Live Control Calibration	Log File Skins About	0.20 _{ppm}
	48 49 50	51 52
Show Alarm	14:00 17 August 2010 Start Stop Clear Start Log St	top Log Show Live Open Save Mem

Release mouse key and the enclosed section will be displayed.

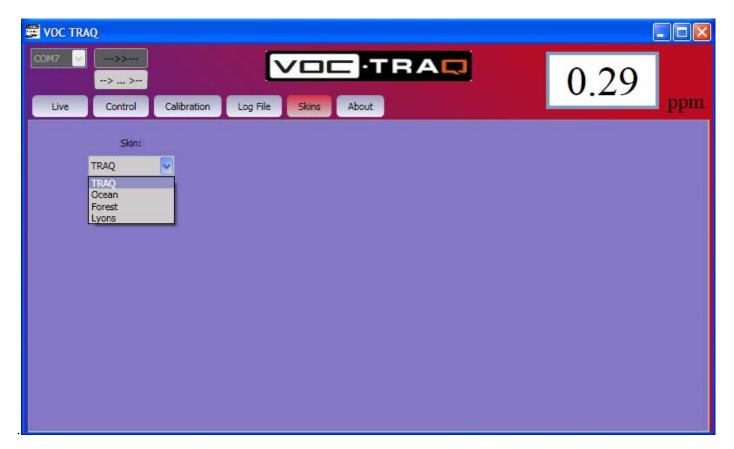
Right click and select "Help". The display "Help" menu will appear.

🏶 About DynamicDataDisplay 🛛 🔀
Dynamic Data Display is in «Auto Fit» mode by default. Using mouse with Shift or Ctrl pressed down you can manually select an area to display.
Mouse wheel, + and - keys - Zoom in/Zoom out
Mouse drag or \leftarrow \downarrow \rightarrow \rightarrow Pan the display area
Ctrl + Mouse drag — Select an area to zoom in
Shift + Mouse drag — Select an area with current
aspect ratio Home — Fit to view
F11 - Copy screenshot of chart to clipboard
Ctrl + S - Save screenshot of chart
http://dynamicdatadisplay.codeplex.com/

These commands give the user control over the graphic display.

2.9 Skins

This pull down menu allows the selection of different color schemes as desired.



2.10 About



Section 3 3.1 VOC-TRAQ Specifications

Sensor: Baseline®-MOCON®, Inc. piD-TECH® plus photoionization detector with 10.6eV Lamp Available Dynamic Ranges (Isobutylene): Part #042-963 0.02 to 20ppm (Silver piD-TECH[®] plus) Part #042-962 0.2 to 200ppm (Bronze piD-TECH® plus) Part #042-961 2.0 to 2000ppm (Black piD-TECH® plus) Target Gases: VOCs and other gases with Ionization Potential <10.6 eV - See Appendix I for an Ionization Potential list. Operating Temperature: -4 to 104 °F (-20 to 40 °C) Operating Humidity: 0 - 90% RH, non-condensing Response Time: T90 10 seconds typical Accuracy: +/- 3% of reading, w/ constant temperature and pressure Dimensions: 1" dia x 3.6" H (2.5cm dia x 9.1cm H) Weight: 1.9 oz (54 g) Internal Memory: 2 Mb EEPROM Memory with programmable sample frequency Output: USB Power: Powered from USB (5.0 VDC, 40 mA) or power supply Calibration: Software controlled Computer Requirements: Windows XP/Vista/7 PC or equivalent via USB Lamp Energy: 10.6 eV Humidity Response: < 1ppm @ 90% r.h. Humidity Quenching Effect: < 15% @ 90% r.h. Serviceable Parts: Lamp, detector cell, filters Warranty Period: 18 months

Section 4 Performance and Maintenance

4.1 Span Drift

A sensor's response to gases may change with time. The common term for this is "Span Drift". The main reason for this drift is typically contamination of the lamp's window.

If the sensor is being used for ambient air applications or applications involving samples containing heavy compounds and/or particles, the lamp window will get contaminated. The rate of the window contamination is a function of the sample gas condition, i.e. how badly it is contaminated with chemicals and particles. Contamination of the lamp window can cause partial UV light blocking, which in turn will reduce the detector's sensitivity. In this case, more frequent calibration is needed and periodic cleaning of the lamp lens. For lamp cleaning instructions refer to Appendix III.

Most VOC's (e.g. isobutylene, benzene) do not contaminate lens and the drift is very small. Typically, span drift does not exceed 10-15% per month of continuous operation. In favorable conditions over a six month period, span drift may be between 15 and 30%.

However, some compounds (such as silicones) are deposited on the lamp window at a more rapid rate. In those circumstances, span drift may be up to 10-20% over an eight hour period.

Also dirt, dust, cleaners, smoke, perfume, etc. in the air can contaminate the lamp window requiring more frequent cleaning.

4.2 VOC-TRAQ's Life Span

The life span of the VOC-TRAQ is basically unlimited; however, there are several PID components that will periodically need replaced depending on the amount of use and the sample that is applied to the sensor.

The UV lamp has a small irreversible internal degradation over time but is insignificant until after 6000 hours of operation. The lens of the lamp can also become contaminated over time if it is exposed to samples containing heavy compounds and/or particles. Periodic calibration of the sensor will compensate for the lamp degradation. If the sensor is used for measuring low-level contaminations in pure gases, it will last as long as the lamp, i.e. >6000 hours without cleaning the lamp or servicing the PID.

All PID replacement parts including the lamp, cell assembly, and filters are listed in Appendix III.

4.4 Temperature Effect

The VOC-TRAQ has a normal operating range from -20°C to 40°C. It will operate safely up to 60°C without damage, however the performance of the VOC-TRAQ is not guaranteed at this elevated temperature. Ambient temperature changes do have an effect on the PID performance. The VOC-TRAQ contains temperature correction to adjust the output to compensate for temperature variations. The variance over the specified temperature range is typically less than plus or minus 5%.

4.5 Response Factors

The ratio between the sensitivity of Isobutylene to that of a target compound is called a Response Factor (RF). For example, VOC-TRAQ has a typical sensitivity of 1mV/ppm for Isobutylene and 2 mV/ppm for Benzene. That means that Benzene's RF is equal to 0.5. Response factors vary to some extent from one PID detector design to another. The response factors are available from various reliable literature sources.

The response factor table (Appendix II) allows the user to measure concentration of various gases without actually calibrating the sensor with the target gas. The following facts and guidelines should be kept in mind while using the response factor table:

- 1. All response Factors were measured in laboratory conditions, with Isobutylene as a reference compound and dry air as a balance gas.
- 2. The actual values of Response Factors may vary in customer's application, depending on the measurement conditions (sample humidity, background gas, lamp condition).
- 3. Response Factors should be used for the approximate measurements, when calibration with the actual target compound is not feasible.
- 4. For the best accuracy, the instrument should be calibrated with the target compound, under the application's conditions.
- 5. Certain gases although they have a response factor tend to be unstable and can cause a photo-chemical reaction in the PID detector. This reaction can cause some unpredictable results. An example of this is NH3 (Ammonia).

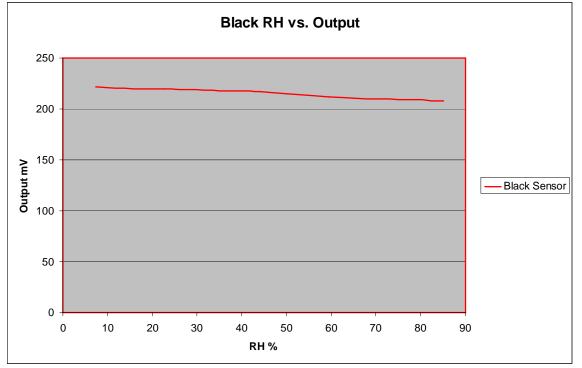
4.6 Response Time (T90/T10)

The time it takes for the signal to go from 0% to 90% of the target gas applied is referred to as the T90 response time and from 100% to 10% is called the T10 response time. The VOC-TRAQ response time is 10 seconds typical and less than 20 seconds maximum. Note that the response time is based on the response of the VOC-TRAQ and not the sample delivery system.

4.7 Humidity Effects

The VOC-TRAQ Humidity Response to Moisture pure Hydrocarbon Free (HCF) air is applied to the sensor, with some humidity present in the sample. The maximum expected shift does not exceed ± 1.0 ppm (Isobutylene). For improving the accuracy of low level measurements, it is recommended to zero the VOC-TRAQ at the same level of relative humidity (RH) as expected in the sample.

The VOC-TRAQ has a relatively flat response to humid environments (see graph below). NOTE: Sudden changes in humidity can cause a temporary shift in output. Allow up to 15 minutes for the VOC TRAQ to acclimate when sudden humidity changes occur.



4.8 VOC-TRAQ Maintenance

The VOC-TRAQ's rugged, durable design provides for trouble-free operation over the course of its lifetime. However, like all photo-ionization sensors, maintenance may be required.

In a polluted environment, window contamination can degrade the sensor's performance. One indication of this problem is a loss of sensitivity. Another possible effect is more noise in a VOC-TRAQ that was properly calibrated. The sensor is still useful with a lower sensitivity but for best results clean the lamp window. For lamp cleaning instructions refer to appendix III

4.9 VOC-TRAQ Disassembly

The VOC-TRAQ contains a piD-TECH® plus photoionization detector with a 10.6eV Lamp. Periodically the lamp in this sensor needs cleaning. The remove the sensor unscrew the two caps from the VOC-TRAQ probe and remove the PCB assemblies. The PID unplugs from the large rectangular printed circuit board (PCB). Refer to appendix III for instructions on cleaning the lamp window. Reassemble the VOC-TRAQ by installing the rectangular PCB into the base cap. Reinsert the PID into the connector. Screw the tubular mid section on to the base cap and the screw on the top cap. See photo below.

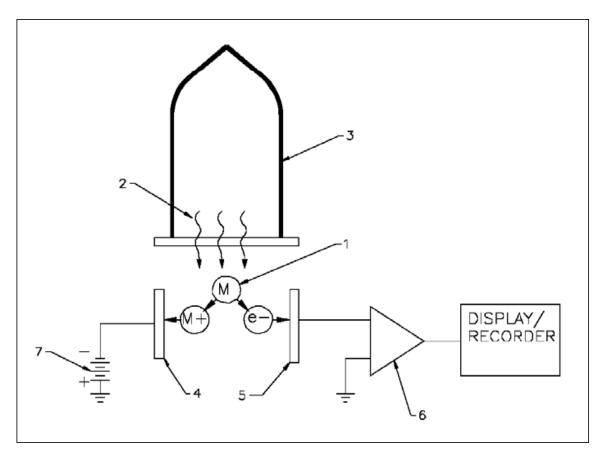


Section 5



5.1 PID Theory of Operation

The VOC-TRAQ is equipped with a Photoionization Detector (PID). The Photoionization Detector (PID) is one of the most widely used gas detection techniques. The main field of PID application is portable instruments for detection of a wide variety of organic compounds and some inorganic gases in ambient air.



A typical PID block diagram is shown above. Molecules of interest (1) are being exposed to high-energy Vacuum Ultra Violet (VUV) radiation (2), generated by the gas discharge lamp (3). As a result, some percentage of these molecules is being ionized, i.e. converted into positively charged ions and negatively charged electrons according to the following equation:

$M + photon --> M^+ + e^-$

To be ionized, the molecule **M** should have its Ionization Potential (IP) smaller than the energy of UV lamp photons (E). As a rule, the bigger the difference is between E an IP, the larger the detector's response. Both E and IP are usually measured in electron-volts (eV). For the Ionization Potentials of various chemicals, refer to Appendix section of this Manual.

The VOC-TRAQ PID is equipped with a 10.6 eV lamp.

The pair of electrodes (4, 5) is located in the ionization volume near the lamp window. The polarizing electrode (4) is connected to the High Voltage DC source (7), the signal electrode (5) is attached to the amplifier (6) input. The electric field, created by these two electrodes, forces both electrons and ions to drift towards their respective electrode, creating a small current. This current is amplified by the amplifier chip and the output analog signal is recorded and/or displayed in digital or analog format. The output signal is proportional to the concentration of ionizable molecules in detector's chamber and thus serves as a measure of concentration. Major air components (N2, O2, and CO2) have ionization potentials greater than the UV lamp and therefore are not detected. For this reason, PID is very useful for detection of a wide range of VOC's (Volatile Organic Compounds) in ambient air, down to the low-ppb concentrations, without interference from air components.

The gaseous sample is delivered to the detector chamber a diffusion process.

Ionization Potentials

Chemical Name	IP (eV)	Benzene	9.25
		Benzenethiol	8.33
Α		Benzonitrile	9.71
2-Aminopyridine	8	Benzotrifluoride	9.68
Acetaldehyde	10.21	Biphenyl	8.27
Acetamide	9.77	Boron oxide	13.5
Acetic acid	10.69	Boron trifluoride	15.56
Acetic anhydride	10	Bromine	10.54
Acetone	9.69	Bromobenzene	8.98
Acetonitrile	12.2	Bromochloromethane	10.77
Acetophenone	9.27	Bromoform	10.48
Acetyl bromide	10.55	Butane	10.63
Acetyl chloride	11.02	Butyl mercaptan	9.15
Acetylene	11.41	cis-2-Butene	9.13
Acrolein	10.1	m-Bromotoluene	8.81
Acrylamide	9.5	n-Butyl acetate	10.01
Acrylonitrile	10.91	n-Butyl alcohol	10.04
Allyl alcohol	9.67	n-Butyl amine	8.71
Allyl chloride	9.9	n-Butyl benzene	8.69
*Ammonia	10.2	n-Butyl formate	10.5
Aniline	7.7	n-Butyraldehyde	9.86
Anisidine	7.44	n-Butyric acid	10.16
Anisole	8.22	n-Butyronitrile	11.67
Arsine	9.89	o-Bromotoluene	8.79
		p-Bromotoluene	8.67
В		p-tert-Butyltoluene	8.28
1,3-Butadiene (butadiene)	9.07	s-Butyl amine	8.7
1-Bromo-2-chloroethane	10.63	s-Butyl benzene	8.68
1-Bromo-2-methylpropane	10.09	sec-Butyl acetate	9.91
1-Bromo-4-fluorobenzene	8.99	t-Butyl amine	8.64
1-Bromobutane	10.13	t-Butyl benzene	8.68
1-Bromopentane	10.1	trans-2-Butene	9.13
1-Bromopropane	10.18		
1-Bromopropene	9.3	С	
1-Butanethiol	9.14	1-Chloro-2-methylpropane	10.66
1-Butene	9.58	1-Chloro-3-fluorobenzene	9.21
1-Butyne	10.18	1-Chlorobutane	10.67
2,3-Butadione	9.23	1-Chloropropane	10.82
2-Bromo-2-methylpropane	9.89	2-Chloro-2-methylpropane	10.61
2-Bromobutane	9.98	2-Chlorobutane	10.65
2-Bromopropane	10.08	2-Chloropropane	10.78
2-Bromothiophene	8.63	2-Chlorothiophene	8.68
2-Butanone (MEK)	9.54	3-Chloropropene	10.04
3-Bromopropene	9.7	Camphor	8.76
3-Butene nitrile	10.39	Carbon dioxide	13.79
Benzaldehyde	9.53	Carbon disulfide	10.07
		Carbon monoxide	14.01
		Carbon tetrachloride	11.47
		Chlorine	11.48
* Contain gagag tand to be unstable of	and oon oonoo		40.00

* Certain gases tend to be unstable and can caues a photo-chemical reaction in the PID detector

Chlorine dioxide

Chlorine trifluoride

10.36

VOC-TRAQUser's Manual

		rippen	anx 1. Iomzation I otential
Chloroacetaldehyde	10.61	Diethylether	9.53
a-Chloroacetophenone	9.44	Diethyl ketone	9.32
Chlorobenzene	9.07	Diethylsulfide	8.43
Chlorobromomethane	10.77	Diethyl sulfite	9.68
Chlorofluoromethane (Freon 22)	12.45	Difluorodibromomethane	11.07
Chloroform	11.37	Dihydropyran	8.34
Chlorotrifluoromethane (Freon 13)	12.91	Diiodomethane	9.34
Chrysene	7.59	Diisopropylamine	7.73
Cresol	8.14	Dimethoxymethane (methylal)	10
Crotonaldehyde	9.73	Dimethyl amine	8.24
Cumene (isopropyl benzene)	8.75	Dimethylether	10
Cyanogen	13.8	Dimethyl sulfide	8.69
Cyclohexane	9.8	Dimethylaniline	7.13
Cyclohexanol	9.75	Dimethylformamide	9.18
Cyclohexanone	9.14	Dimethylphthalate	9.64
Cyclohexene	8.95	Dinitrobenzene	10.71
Cyclo-octatetraene	7.99	Dioxane	9.19
Cyclopentadiene	8.56	Diphenyl	7.95
Cyclopentane	10.53	Dipropylamine	7.84
Cyclopentanone	9.26	Dipropyl sulfide	8.3
Cyclopentene	9.01	Durene	8.03
Cyclopropane	10.06	m-Dichlorobenzene	9.12
m-Chlorotoluene	8.83	N,N-Diethyl acetamide	8.6
o-Chlorotoluene	8.83	N,N-Diethylformamide	8.89
p-Chlorotoluene	8.7	N,N-Dimethyl acetamide	8.81
		N,N-Dimethyl formamide	9.12
D		o-Dichlorobenzene	9.06
1,1-Dibromoethane	10.19	p-Dichlorobenzene	8.95
1,1-Dichloroethane	11.12	p-Dioxane	9.13
1,1-Dimethoxyethane	9.65	trans-Dichloroethene	9.66
1,1-Dimethylhydrazine	7.28		
1,2-Dibromoethene	9.45	E	
1,2-Dichloro-1,1,2,2-tetrafluoroethane		Epichlorohydrin	10.2
(Freon 114)	12.2	Ethane	11.65
1,2-Dichloroethane	11.12	Ethanethiol (ethyl mercaptan)	9.29
1,2-Dichloropropane	10.87	Ethanolamine	8.96
1,3-Dibromopropane	10.07	Ethene	10.52
1,3-Dichloropropane	10.85	Ethyl acetate	10.11
2,2-Dimethyl butane	10.06	Ethyl alcohol	10.48
2,2-Dimethylpropane	10.35	Ethyl amine	8.86
2,3-Dichloropropene	9.82	Ethylbenzene	8.76
2,3-Dimethyl butane	10.02	Ethyl bromide	10.29
3,3-Dimethyl butanone	9.17	Ethyl chloride (chloroethane)	10.98
cis-Dichloroethene	9.65	Ethyl disulfide	8.27
Decaborane	9.88	Ethylene	10.5
Diazomethane	9	Ethyl ether	9.51
Diborane	12	Ethyl formate	10.61
Dibromochloromethane	10.59	Ethyl iodide	9.33
Dibromodifluoromethane	11.07	Ethyl isothiocyanate	9.14
Dibromomethane	10.49	Ethyl mercaptan	9.29
Dibutylamine	7.69	Ethyl methyl sulfide	8.55
Dichlorodifluoromethane (Freon 12)	12.31	Ethyl nitrate	11.22
Dichlorofluoromethane	12.39	Ethylpropionate	10
Dichloromethane	11.35	Ethyl thiocyanate	9.89
Diethoxymethane	9.7	Ethylene chlorohydrin	10.52
Diethylamine	8.01	Ethylene diamine	8.6

Baseline - MOCON, Inc.

Appendix I: Ionization Potentials

Ethylene dibromide	10.37	Isobutyl acetate	9.97
Ethylene dichloride	11.05	Isobutyl alcohol	10.12
Ethylene oxide	10.57	Isobutylamine	8.7
Ethylenelmine	9.2	Isobutyl formate	10.46
Ethynylbenzene	8.82	Isobutyraldehyde	9.74
		Isobutyric acid	10.02
F		Isopentane	10.32
2-Furaldehyde	9.21	Isophorone	9.07
Fluorine	15.7	Isoprene	8.85
Fluorobenzene	9.2	Isopropylacetate	9.99
Formaldehyde	10.87	Isopropylalcohol	10.16
Formamide	10.25	Isopropylamine	8.72
Formic acid	11.05	Isopropylbenzene	8.69
Freon 11 (trichlorofluoromethane)	11.77	Isopropylether	9.2
Freon 112 (1,1,2,2-tetrachloro-1,2-difluoroetha	,	Isovaleraldehyde	9.71
Freon 113 (1,1,2-trichloro-1,2,2-trifluororethand	,	m-lodotoluene	8.61
Freon 114 (1,2-dichloro-1,1,2,2-tetrafluoroetha	,	o-lodotoluene	8.62
Freon 12 (dichlorodifluoromethane)	12.31	p-lodotoluene	8.5
Freon 13 (chlorotrifluoromethane)	12.91		
Freon 22 (chlorofluoromethane)	12.45	К	
Furan	8.89	Ketene	9.61
Furfural	9.21	_	
m-Fluorotoluene	8.92	L	
o-Fluorophenol	8.66	2,3-Lutidine	8.85
o-Fluorotoluene	8.92	2,4-Lutidine	8.85
p-Fluorotoluene	8.79	2,6-Lutidine	8.85
н		Μ	
1-Hexene	9.46	2-Methylfuran	8.39
2-Heptanone	9.33	2-Methyl napthalene	7.96
2-Hexanone	9.35	1-Methyl napthalene	7.96
Heptane	10.08	2-Methylpropene	9.23
Hexachloroethane	11.1	2-Methyl-1-butene	9.12
Hexane	10.18	2-Methylpentane	10.12
Hydrazine	8.1	3-Methyl-1-butene	9.51
Hydrogen	15.43	3-Methyl-2-butene	8.67
Hydrogen bromide	11.62	3-Methylpentane	10.08
Hydrogen chloride	12.74	4-Methylcyclohexene	8.91
Hydrogen cyanide	13.91	Maleicanhydride	10.8
Hydrogenfluoride	15.77	Mesityl oxide	9.08
Hydrogen iodide	10.38	Mesitylene	8.4
Hydrogen selenide	9.88	Methane	12.98
Hydrogen sulfide	10.46	Methanethiol (methyl mercaptan)	9.44
Hydrogen telluride	9.14	Methyl acetate	10.27
Hydroquinone	7.95	Methyl acetylene	10.37
		Methyl acrylate	9.9
I		Methyl alcohol	10.85
1-lodo-2-methylpropane	9.18	Methyl amine	8.97
1-lodobutane	9.21	Methyl bromide	10.54
1-lodopentane	9.19	Methyl butyl ketone	9.34
1-lodopropane	9.26	Methyl butyrate	10.07
2-lodobutane	9.09	Methyl cellosolve	9.6
2-lodopropane	9.17	Methyl chloride	11.28
lodine	9.28	Methyl chloroform (1,1,1-trichloroethane)	11
lodobenzene	8.73	Methyl disulfide	8.46
Isobutane (Isobutylene)	9.4	Methyl ethyl ketone	9.53

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VOC-TRAQUSER SIManual		Appendix 1: Iomzau	ion Potential
Methyl formate	10.82	Phenyl ether (diphenyl oxide)	8.82
Methyliodide	9.54	Phenylhydrazine	7.64
Methyl isobutyl ketone	9.3	Phenyl isocyanate	8.77
Methyl isobutyrate	9.98	Phenyl isothiocyanate	8.52
Methyl isocyanate	10.67	Phenylene diamine	6.89
Methyl isopropyl ketone	9.32	Phosgene	11.77
Methyl isothiocyanate	9.25	Phosphine	9.87
Methyl mercaptan	9.44	Phosphorus trichloride	9.91
Methyl methacrylate	9.7	Phthalic anhydride	10
Methyl propionate	10.15	Propane	11.07
Methyl propyl ketone	9.39	Propargylalcohol	10.51
a -Methyl styrene	8.35 10.07	Propiolactone	9.7 9.98
Methyl thiocyanate	10.07	Propionaldehyde Propionia acid	9.98 10.24
Methylal (dimethoxymethane) Methylcyclohexane	9.85	Propionic acid Propionitrile	10.24
Methylene chloride	11.32	Propyl acetate	10.04
Methyl-n-amyl ketone	9.3	Propylalcohol	10.04
Monomethylaniline	7.32	Propylamine	8.78
Monomethyl hydrazine	7.67	Propylbenzene	8.72
Morpholine	8.2	Propylether	9.27
n-Methyl acetamide	8.9	Propylformate	10.54
	0.0	Propylene	9.73
Ν		Propylene dichloride	10.87
1-Nitropropane	10.88	Propylene imine	9
2-Nitropropane	10.71	Propylene oxide	10.22
Naphthalene	8.12	Propyne	10.36
Nickel carbonyl	8.27	Pyridine	9.32
Nitric oxide, (NO)	9.25	Pyrrole	8.2
Nitrobenzene	9.92		
Nitroethane	10.88	Q	
Nitrogen	15.58	Quinone	10.04
Nitrogen dioxide	9.78		
Nitrogen trifluoride	12.97	S	
Nitromethane	11.08	Stibine	9.51
Nitrotoluene	9.45	Styrene	8.47
p-Nitrochloro benzene	9.96	Sulfurdioxide	12.3
		Sulfur hexafluoride	15.33
0		Sulfurmonochloride	9.66
Octane	9.82	Sulfurylfluoride	13
Oxygen	12.08	_	
Ozone	12.08	T	
_		o-Terphenyls	7.78
P		1,1,2,2-Tetrachloro-1,2-difluoroethane (Freon 1	
1-Pentene	9.5	1,1,1-Trichloroethane	11
1-Propanethiol	9.2	1,1,2-Trichloro-1,2,2-trifluoroethane (Freon 113	,
2,4-Pentanedione	8.87	2,2,4-Trimethylpentane	9.86
2-Pentanone	9.38	o-Toluidine	7.44
2-Picoline	9.02	Tetrachloroethane	11.62
3-Picoline	9.02	Tetrachloroethene	9.32
4-Picoline	9.04	Tetrachloromethane	11.47
n-Propyl nitrate	11.07	Tetrahydrofuran	9.54
Pentaborane	10.4 10.25	Tetrahydropyran Thiolocotic acid	9.25
Pentane	10.35	Thiolacetic acid	10
Perchloroethylene Pheneloic	9.32 8.18	Thiophene Toluene	8.86 8.82
Phenol	8.5	Tribromoethene	0.02 9.27
	0.0		5.21

Appendix I: Ionization Potentials

Tribromofluoromethane	10.67
Tribromomethane	10.51
Trichloroethene	9.45
Trichloroethylene	9.47
Trichlorofluoromethane (Freon 11)	11.77
Trichloromethane	11.42
Triethylamine	7.5
Trifluoromonobromo-methane	11.4
Trimethylamine	7.82
Tripropylamine	7.23

V

•	
o-Vinyl toluene	8.2
Valeraldehyde	9.82
Valeric acid	10.12
Vinyl acetate	9.19
Vinylbromide	9.8
Vinyl chloride	10
Vinyl methyl ether	8.93

W Water	12.59
X	7.05
2,4-Xylidine	7.65
m-Xylene	8.56

m-Xylene	8.56
o-Xylene	8.56
p-Xylene	8.45

Response Factors

1,2,3-trimethylbenzene	0.49	ethylene glycol	15.7
1,2,4-trimethylbenzene	0.43	ethylene oxide	19.5
1,2-dibromoethane	11.7	gasoline	1.1
1,2-dichlorobenzene	0.5	heptane	2.5
1,2-dichloroethane (11.7 lamp)	0.5	hydrazine	2.6
1,3,5-trimethylbenzene	0.34	hydrogen sulfide	3.2
1,4-dioxane	1.4	isoamyl acetate	1.8
1-butanol	3.4	isobutanol	4.7
1-methoxy-2-propanol	1.4	isobutyl acetate	2.6
1-propanol	5.7	isobutylene	1
2-butoxyethanol	1.3	isooctane	1.3
2-methoxyethanol	2.5	isopentane	8
2-pentanone	0.78	isophorone	0.74
2-picoline	0.57	isoprene (2-methyl-1,3-butadiene)	0.6
3-picoline	0.9	isopropanol	5.6
, 4-hydroxy-4-methyl-2-pentanone	0.55	isopropyl acetate	2.6
4-methylbenzylalcohol	0.8	isopropylether	0.8
acetaldehyde	10.8	isopropylamine	0.9
acetic acid	11	Jet A fuel	0.4
acetone	1.2	JP-5 fuel	0.48
acetophenone	0.59	JP-8 fuel	0.48
acrolein	3.9	mesityl oxide	0.47
allyl alcohol	2.5	methanol (11.7 lamp)	2.5
ammonia	9.4	methyl acetate	7
amylacetate	3.5	methyl acetoacetate	1.1
arsine	2.6	methyl acrylate	3.4
benzene	0.53	methylbenzoate	0.93
bromoform	2.3	methyl ethyl ketone	0.9
bromomethane	1.8	methyl isobutyl ketone	1.1
butadiene	0.69	methylmercaptan	0.6
butyl acetate	2.4	methyl methacrylate	1.5
carbon disulfide	1.2	methyl tert-butyl ether	0.86
chlorobenzene	0.4	methylamine	1.2
cumene (isopropylbenzene)	0.54	methylene chloride (11.7 lamp)	0.85
cyclohexane	1.5	m-xylene	0.53
cyclohexanone	0.82	naphtalene	0.37
decane	1.6	n,n-dimethylacetamide	0.73
diethylamine	1	n,n-dimethylformamide	0.8
dimethoxymethane	11.3	n-hexane	4.5
dimethyl disulfide	0.3	nitric oxide	7.2
diesel fuel #1	0.9	n-nonane	1.6
diesel fuel #2	0.75	nitrogen dioxide (11.7 lamp)	10
epichlorhydrin	7.6	n-pentane	9.7
ethanol athul apotato	10	n-propylacetate	3.1 2.2
ethyl acetate	4.2	octane	
ethyl acetoacetate	0.9 2.3	o-xylene	0.54 1
ethyl acrylate ethyl ether (diethyl ether)	2.3 1.2	phenol phosphine	2.8
ethylmercaptan	0.6	pinene, alpha	0.4
ethylbenzene	0.51	pinene, beta	0.4
ethylene	10.1	propionaldehyde (propanal)	14.8
	10.1	propronalaonyao (propanal)	0.71

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propylene	1.3
propylene oxide	6.5
p-xylene	0.5
pyridine	0.79
quinoline	0.72
styrene	0.4
tert-butyl alcohol	3.4
tert-butyl mercaptan	0.55
tert-butylamine	0.71
tetrachloroethylene	0.56
tetrahydrofuran	1.6
thiophene	0.47
toluene	0.53
trans-1,2-Dichloroethene	0.45
trichloroethylene	0.5
trimethylamine	0.83
turpentine - crude sulfite	1
turpentine - pure gum	0.45
vinylacetate	1.3
vinylbromide	0.4
vinylchloride	1.8
vinylcyclohexane (VCH)	0.54
vinylidene chloride (1,1-DCE)	0.8

* Certain gases tend to be unstable and can cause a photo-chemical reaction in the PID detector

Servicable Items and Instructions

All **piD-TECH** ® **plus** Sensors contain six user replaceable components:



FilterCap(P/N037-581)

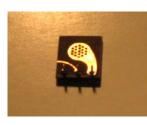


Spacer (P/N042-078)



Filter Media#1 (P/N038-083)

Filter Media#2 (P/N037-591)



 $Cell Assembly (P\!/\!N\,042\text{-}216)$



10.6eV Lamp (P/N038-566)

Warning

All maintenece procedures must be performed on a clean surface using clean tools. Avoid touching the lamp's window as well as the metalized portion of the Cell Assembly with your bare fingers. Fingerprints left on these parts may adversely affect the sensors operation.Latex gloves are preferred, but if they are not used, your hands must be clean and free of oils, lotions, etc. It is acceptable to hold the lamp by its glass body or by the edges of the window.

Tools Required

- X-Acto Knife (preferred) or Small Slotted Screwdriver
- Fine-Tipped Tweezers
- Latex Gloves (Optional)

Maintenance Kit List

The following maintenance kits are offered:

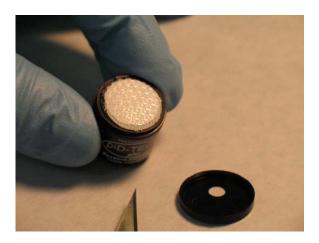
Description	Part No
Dry Lamp Cleaning Kit	042-246
Replacement Filter Set	042-205
Replacement Filter Set w/Cap	036-211

Disassembly

- 1. Power down the instrument according to the User's Manual and remove the sensor from the instrument.
- 2. Remove the Filter Cap by applying slight upward pressure with the tip of a screwdriver or X-Acto blade just below the hole in the cap and between the cap and housing, it will pop off.



3. With fine-tipped tweezers, remove both the Filter Media and set aside.



4. Using the X-Acto blade, remove the spacer and set aside.

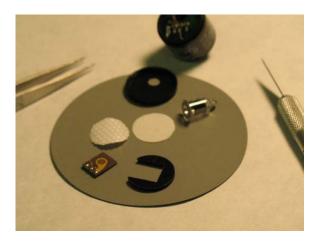


5. With fine-tipped tweezers, carefully remove the Cell Assembly by prying under the Cell's edge where connector pins are located.



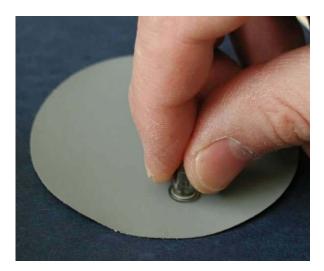
6. With fine-tipped tweezers, grasp the lamp by placing the tips in the housing notch and gently pull it out. Be careful not to scratch the lamp lens or chip edges.





Cleaning the Lamp

Grab the lamp by the cylindrical glass body and clean the window by rubbing it against the Polishing Pad. Use a circular motion and try to keep the window surface flat relative to the pad. Five seconds of rubbing will be enough in most cases. Another indication of cleaning completeness is that you have used about 1/6th of the pads surface during the procedure.

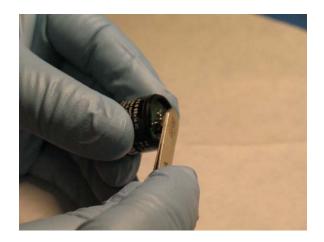


Reassembly

1. Install the lamp into the sensor, making sure that the lamps metalized pads are aligned with the corresponding excitation springs inside the lamp cavity.



2. With the end of the clean tweezers, or the clean blade of a screwdriver, press down firmly. Be careful not to scratch the surface of the lamp.



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3. Using fine-tipped tweezers, install the cell assembly. Align the pins with the corresponding sockets on the sensor and push down on the end with the pins. Make sure the cell assembly is flush with the lamp window.



4. Place the spacer around the cell assembly.



5. Place the Filter Media over the Cell Assembly centered on top of the sensor. Make sure the filters are installed in the correct order. Filter Media #2 first, then Filter Media #1 on top, with the shiny side up.

Appendix III: Servicable Items and Instructions



6. Alight the Cap Key with the notch on the housing. Starting at the side opposite the notch, press down until the Filter Cap snaps on to the housing. If the Cap Key is incorrectly aligned, there will be a noticeable bulge on the side of the cap.

